

## 3-(2,4-Dichlorophenyl)-3-hydroxy-1-(3-nitrophenyl)propan-1-one

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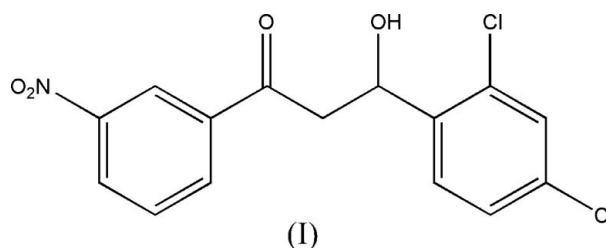
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## Key indicators

Single-crystal X-ray study  
 $T = 298$  K  
Mean  $\sigma(\text{C}-\text{C}) = 0.006$  Å  
 $R$  factor = 0.063  
 $wR$  factor = 0.153  
Data-to-parameter ratio = 14.5For details of how these key indicators were  
automatically derived from the article, see  
<http://journals.iucr.org/e>.The title compound,  $\text{C}_{15}\text{H}_{11}\text{Cl}_2\text{NO}_4$ , has been synthesized and characterized by single-crystal X-ray diffraction analysis. All bond lengths and angles are within normal ranges. The dihedral angle between the two benzene rings is  $77.2(2)^\circ$ .Received 2 March 2007  
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## Comment

Recently, we have reported a few chalcone derivatives (Qiu, Fang *et al.*, 2006; Qiu, Liu *et al.*, 2006; Qiu, Luo *et al.*, 2006). As an extension of our work on the structural characterization of chalcone derivatives, we report here the structure of the title compound, (I). A view of (I) with the atom-numbering scheme is given in Fig. 1. In (I), all bond lengths and angles are within normal ranges (Allen *et al.*, 1987). The dihedral angle between the two benzene rings is  $77.2(2)^\circ$ . The crystal structure is stabilized by hydrogen bonds (Table 1).

## Experimental

The reagents were commercial products and were used without further purification. An aqueous solution of potassium hydroxide (5%, 1 ml) was added with stirring overnight to a solution of 2,4-dichlorobenzaldehyde (1 mmol, 0.18 g) and 1-(3-nitrophenyl)ethanone (1 mmol, 0.17 g) in ethanol (15 ml) at room temperature. The reaction mixture was then poured into ice and neutralized with hydrochloric acid (5%). A yellow solid was obtained from ethanol. The solid was dissolved in acetone (12 ml) and stirred for about 10 min to give a clear yellow solution. After keeping the solution in air for 9 d, yellow block-shaped crystals were formed at the bottom of the vessel on slow evaporation of the solvent. These were collected, washed three times with acetone and dried in a vacuum desiccator using  $\text{CaCl}_2$ . The compound was isolated in 65% yield.

## Crystal data

 $\text{C}_{15}\text{H}_{11}\text{Cl}_2\text{NO}_4$   
 $M_r = 340.15$   
Triclinic,  $P\bar{1}$   
 $a = 7.4440(15)$  Å  
 $b = 8.5460(17)$  Å  
 $c = 12.257(3)$  Å  
 $\alpha = 77.86(3)^\circ$   
 $\beta = 81.28(3)^\circ$  $\gamma = 83.18(3)^\circ$   
 $V = 750.4(3)$  Å<sup>3</sup>  
 $Z = 2$   
Mo  $K\alpha$  radiation  
 $\mu = 0.45$  mm<sup>-1</sup>  
 $T = 298(2)$  K  
 $0.30 \times 0.10 \times 0.10$  mm

Data collection

Bruker SMART APEX area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)  
 $T_{\min} = 0.945$ ,  $T_{\max} = 0.954$

3184 measured reflections  
2945 independent reflections  
1683 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.032$

Refinement

$R[F^2 > 2\sigma(F^2)] = 0.063$   
 $wR(F^2) = 0.153$   
 $S = 1.03$   
2945 reflections  
203 parameters

H atoms treated by a mixture of independent and constrained refinement  
 $\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$   
 $\Delta\rho_{\text{min}} = -0.27 \text{ e \AA}^{-3}$

Table 1

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ ).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$O4-H10\cdots O3^i$	0.97 (5)	1.89 (5)	2.852 (4)	171 (4)
$C8-H7B\cdots O1^{ii}$	0.97	2.56	3.474 (4)	157
$C9-H9\cdots Cl1$	0.98	2.67	3.042 (4)	103
$C12-H12\cdots O4^{iii}$	0.93	2.45	3.275 (4)	148
$C14-H14\cdots O3^{ii}$	0.93	2.52	3.340 (4)	147
$C15-H15\cdots O4$	0.93	2.50	2.816 (4)	100

Symmetry codes: (i)  $-x + 1, -y + 2, -z$ ; (ii)  $x, y + 1, z$ ; (iii)  $x + 1, y, z$ .

H atoms of the CH and CH<sub>2</sub> groups in (I) were positioned geometrically and treated as riding with C–H distances of 0.93–0.98  $\text{\AA}$  and with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$ . The H atom of the OH group was located in a difference Fourier map and refined freely.

Data collection: SMART (Bruker, 1998); cell refinement: SAINT (Bruker, 1998); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine

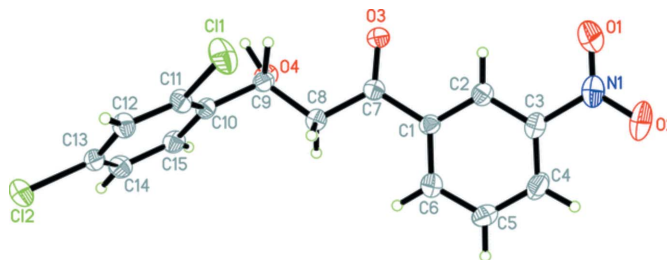


Figure 1

The molecular structure of (I), showing 30% probability displacement ellipsoids and the atom-numbering scheme.

structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL (Sheldrick, 1999); software used to prepare material for publication: SHELXTL.

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